

High Energy Lithium Batteries for PHEV Applications

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Program Overview

Time line

Start Date: October 2013

End Date: March 2016

Status: Completed

Budget

Total Project Funding:

\$ 3,785,088

Cost Share:

\$ 757,018

Funding Received:

\$ 1,433,992 (Envia)

Barriers

- Meeting PHEV power specifications
- Loss of power with cycling
- Cycle and calendar life

Partners







Project Lead











Outline

- **□** Overview & Objectives
- Motivation & Roadmap
- Materials Engineering
 - Compositional Engineering
 - Surface Engineering
- ☐ Diagnostic Studies
 - Structure
 - Interphase/Interface
 - Electrochemical
- ☐ Theoretical Modeling
- ☐ Cell Engineering
- ☐ Conclusions & Acknowledgements









Project Objectives - Relevance

<u>Goals</u>

Develop a high capacity cathode, and $Si-SiO_x$ -C based anode and integrate them and build high capacity (0.25-40 Ah) pouch cells that exceed the ABR minimum target goals for PHEVs

Relevance

 Identifying the root cause and solving the DC-Resistance rise at low SOC's, enabling the use of the high-energy offered by HCMR[™] Lirich cathode materials

Project Tasks

- Material development
- Nanocoating engineering
- Atomistic and cell-level modeling
- Material scale-up
- Large cell development
- Large cell testing

Cell Targets

Characteristics	Unit	PHEV40
Specific Discharge Pulse Power	W/kg	800
Discharge Pulse Power Density	W/L	1600
Specific Regen Pulse Power	W/kg	430
Regen Pulse Power Density	W/L	860
Recharge Rate	NA	C/3
Specific Energy	Wh/kg	200
Energy Density	Wh/L	400
Calendar Life	Years	10+
Cycle Life (at 30° C with C/3 Charge and 1C Discharge rates)	Cycles	5000
Operating Temperature Range	°C	-30 to +52

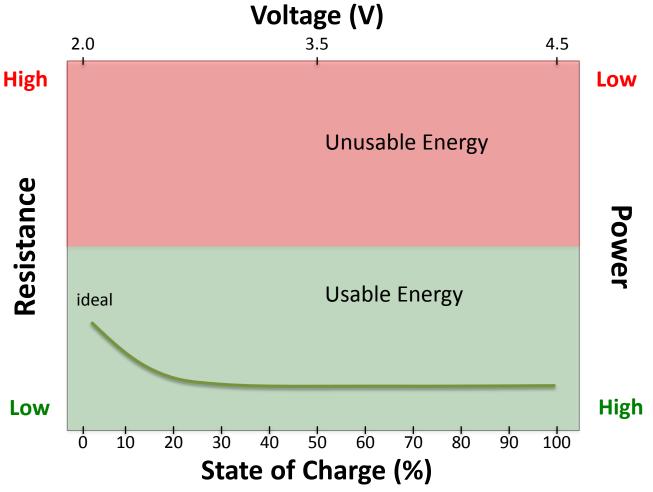








Importance of Cell DC-R/Power for Automotive Range





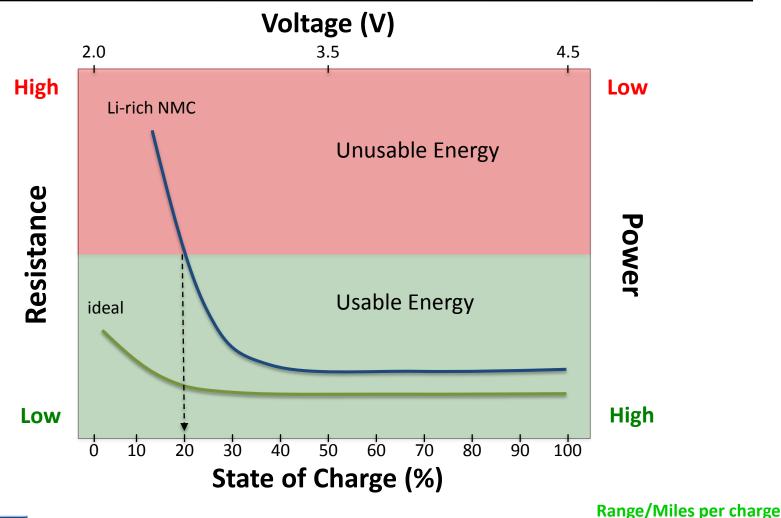




Range/Miles per charge



Importance of Cell DC-R/Power for Automotive Range









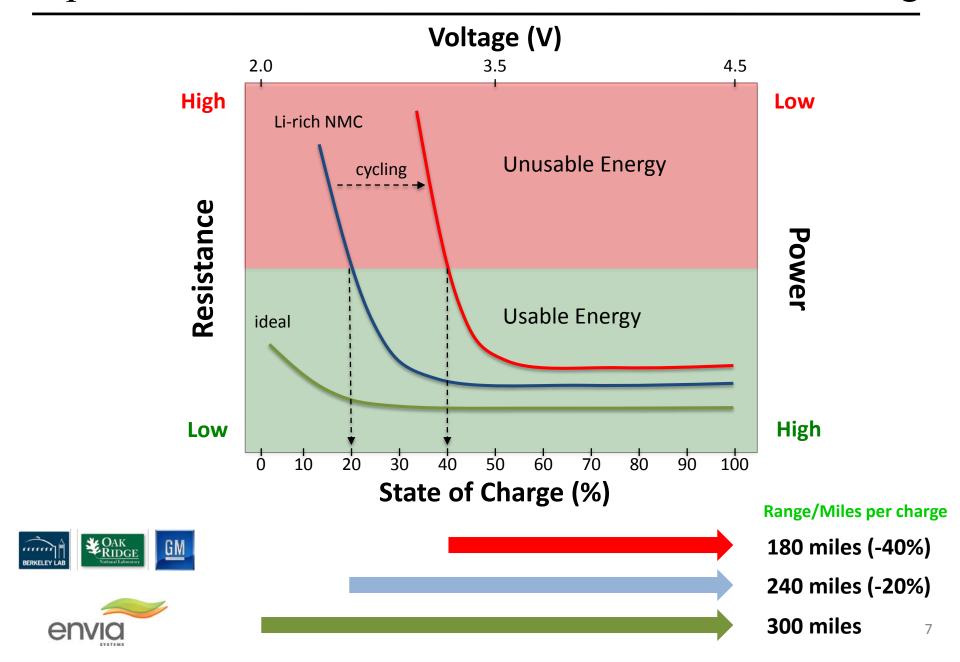




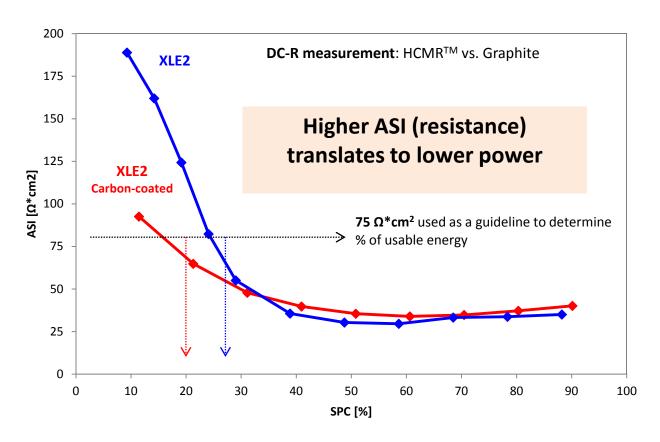
240 miles (-20%)

300 miles

Importance of Cell DC-R/Power for Automotive Range



Motivation: Reduction of DC-R by Surface Coatings



- HCMR[™]-XLE2 material has only ~75% energy extracted from lower resistance regions
- Nevertheless, the same chemistry with a carboncoating increases usable energy from ~75% to >82%

It is possible to increase usable energy by engineering surface coating chemistries and processes!

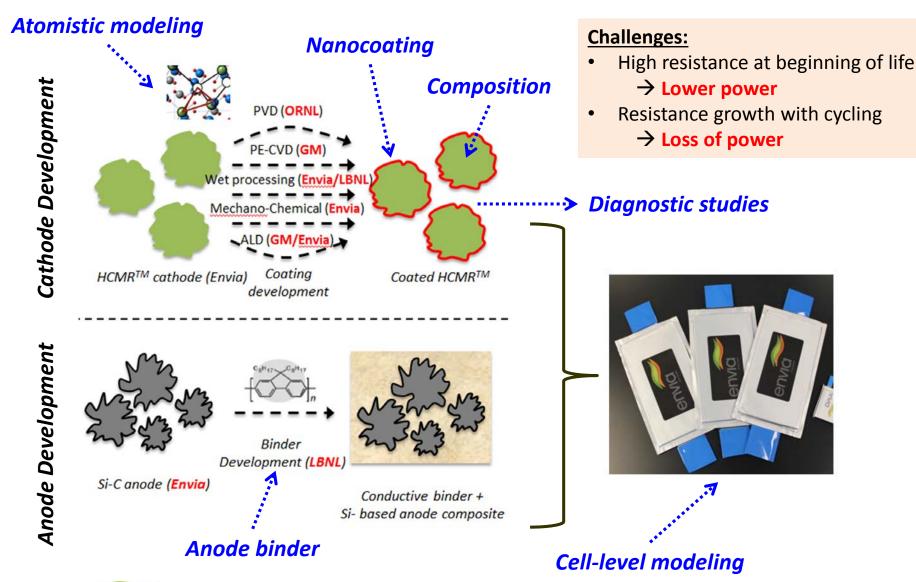








Project Development Roadmap











Project Tasks & Timeline

Task	Sub Tasks	Timeline										
		Q1	Q2	Q3	Q4	Q5	Q6	Q7	Q8	Q9	Q10	Q11
	1. Composition Engineering	◆(XLE)				◆ (XE)						
	2. Surface coating											
	i. LiPON											
Cathode	ii. ALD											
Development	iii. Polymer											
	iv. Carbon							,				
	3. Atomistic Modeling											
	4. Diagnostic studies											
Anode	1. Composition Engineering											
Development	2. Binder Development											
Cell Development	1. Cell Sizing Studies											
	2. Electrochemical Modeling											
	3. 1~50 Ah Cell Builds (Internal)											
	4. 1~50 Ah Final Cell Build											

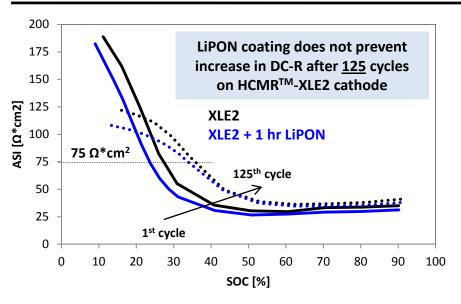


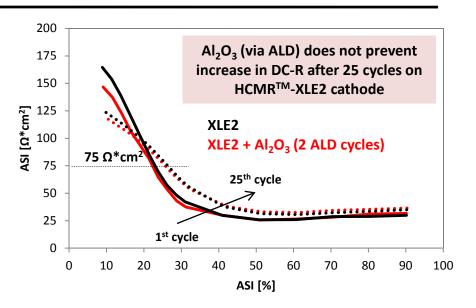


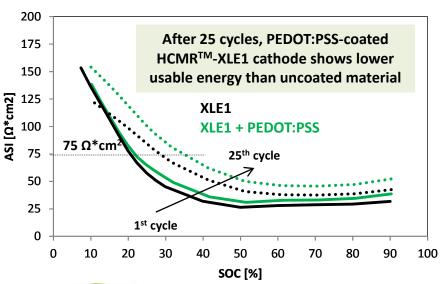




Surface Modification on HCMRTM-XLE Cathode







LiPON, Al₂O₃ (via ALD), and polymer (PEDOT:PSS) coatings on HCMRTM-XLE did not prevent loss of usable energy

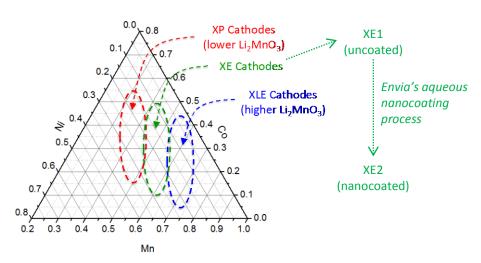


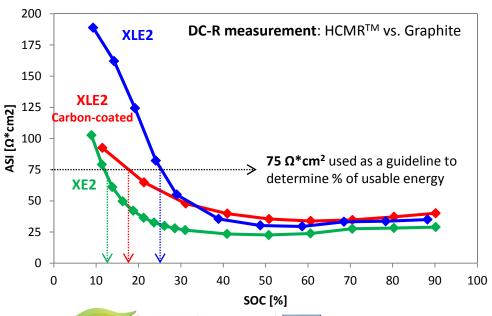


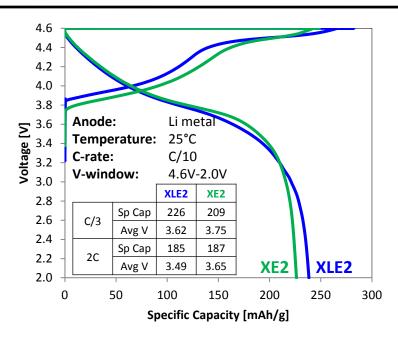




Compositional Engineering







PHEV cells demand high power – The newly developed composition HCMR™-XE is better or on-par in capacity at high rates with higher average voltage than HCMR™-XLE material.

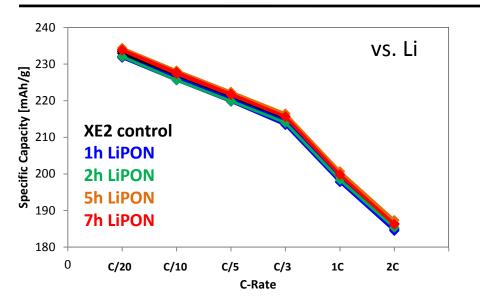


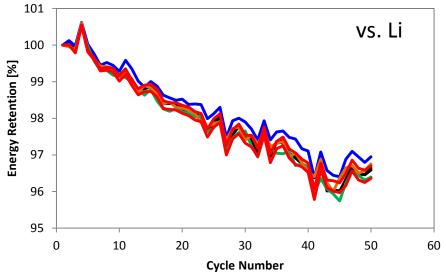






LiPON-Coated XE2: Electrochemical Properties





- LiPON coated XE2 materials perform nearly identically to uncoated
- No improvement in capacity and average voltage signifies no effective enhancement in DC-R

LiPON coating does not improve rate-capability nor cycle-life.

Negative impact on Average Voltage at faster rates.

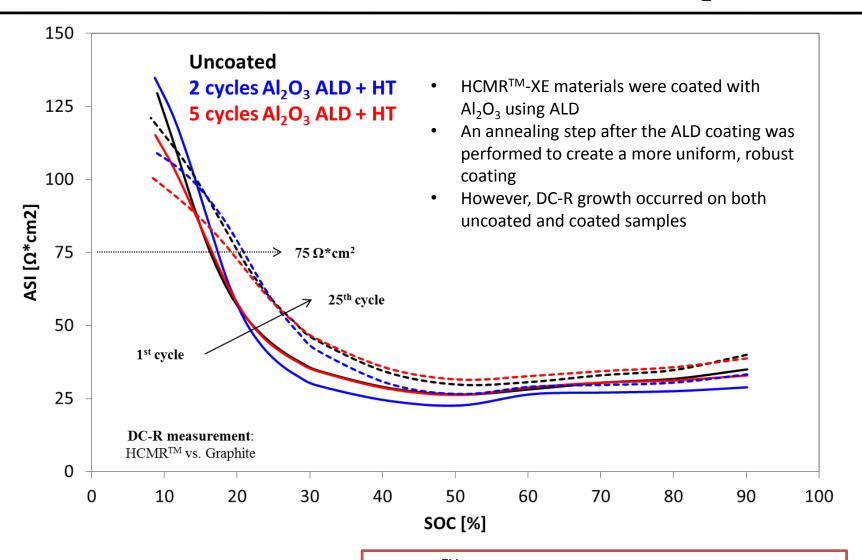








ALD-Coated XE2: Electrochemical Properties







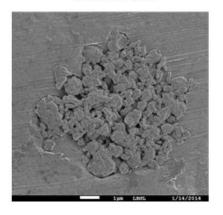




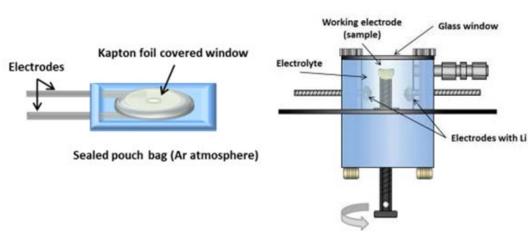
HCMRTM-XE ALD coating studies with High Temperature treatment did not prevent DC-R growth after 25 cycles

Diagnostics Studies

Section of model electrode



In situ XRD cell



In situ Raman cell

STEM – Microscopic Analysis @ NCEM & Super STEM, UK



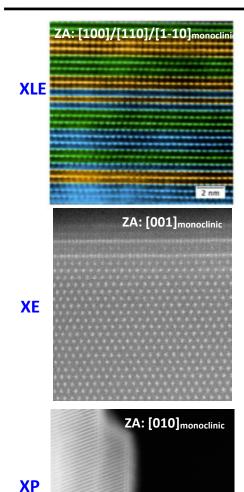




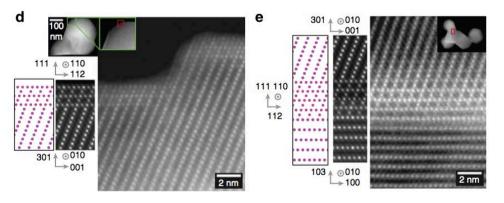




Study of Structure Using HAADF STEM imaging



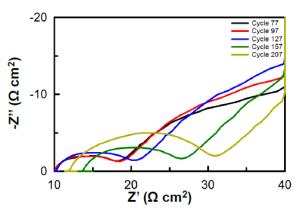
5 nm

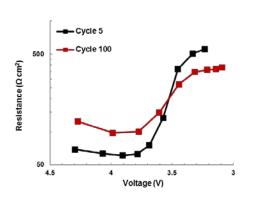


A. K. Shukla et al. Nat Commun, vol. 6, p. 8711, Oct. 2015

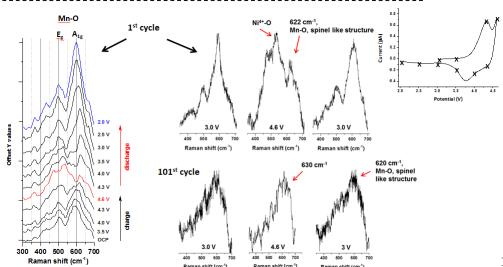
- Bulk structure: Preliminary results obtained from HAADF STEM analysis of XLE, XE and XP materials suggests that as Li and Mn content is decreased, there is an increase in variation of lithium content in the shared column present in monoclinic domains as observed from [100] monoclinic zone axis.
- Surface structure: All three materials exhibited the presence of spinel surface having the same orientation relationship with the bulk structure and also with similar thickness (~2 nm).
- The results indicate that the composition does not have an effect on the thickness of this spinel layer, indicating that the bulk structure might be more responsible in the reduction of DC-R with decrease in Li/M ratio.

Diagnostics – Spectroscopy & Electrochemical





- Li+ diffusivity changes dramatic below 3.8 V within a cycle
- No change of Li+ diffusivity in HCMRTM material with prolonged cycling
- EIS studies show increase in surface film thickness mainly due to loss in particle contact.
- EC analysis of HCMR™-XE and HCMR™-XLE cathodes show that DC-R is caused by surface film buildup at lower states of charge
- Cathode material diffusivity changes due to changes in the lattice parameter during a single cycle.
 - Surface changes to a spinel like structure. Changes in surface transition metal activity.
 - Ni⁴⁺/Ni²⁺ redox reactions observed during 1st cycle
 - No Ni⁴⁺/Ni²⁺ redox reactions observed after 100 cycles



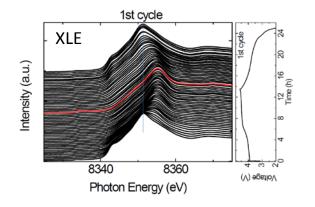


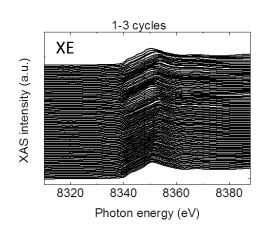


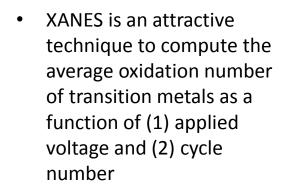


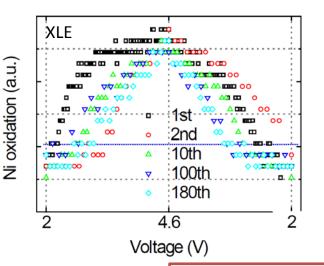


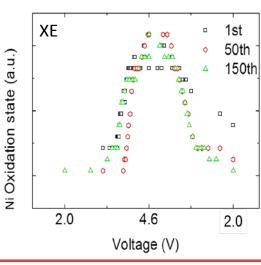
NE-XANES – HCMRTM XLE vs. XE











- Change in Ni oxidation number from 2+ to 4+ when charging the cell to 4.6V
- Behavior at each cycle is more stable for the HCMRTM-XE2 material than HCMRTM-XLE2.

HCMR[™]-XE2 cathode shows less irreversibility with respect to Ni oxidation number profile compared to HCMR[™]-XLE2.



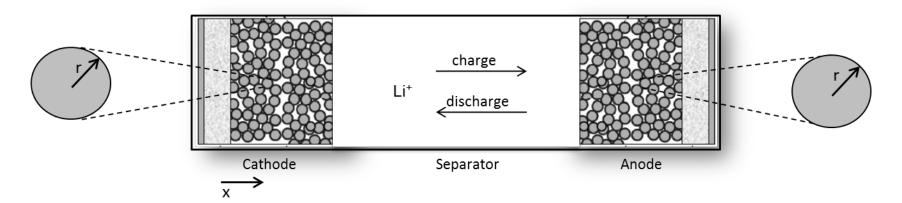






Electrochemical Modeling

Schematic of a single unit cell



- Pseudo-2D model accounting for variation in the field variables along the thickness of the unit cell
- Base model accounts for potential gradients in solid and liquid (Ohm's law), concentration gradients in solid and liquid (Fick's law), and intercalation reaction (Butler-Volmer kinetics)

Model Inputs

- Cell design parameters
- Electrolyte properties from literature
- Fitted parameters:
 - Solid state diffusivity
 - SEI film resistance
 - Reaction kinetics
 - Electrode tortuosity



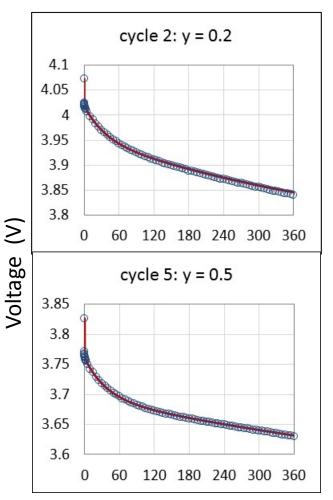


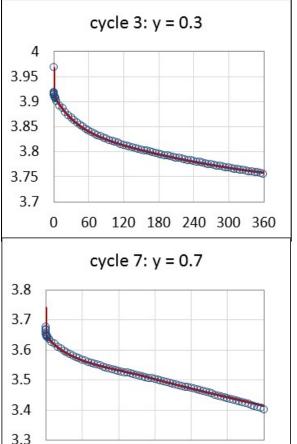


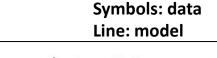


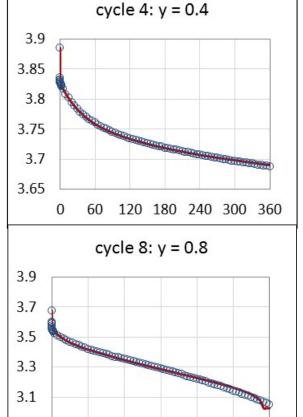
Electrochemical Model Fit (HCMRTM XE1)

Half cell voltage as a function of time









Time (s)









Good agreement between model and experiment

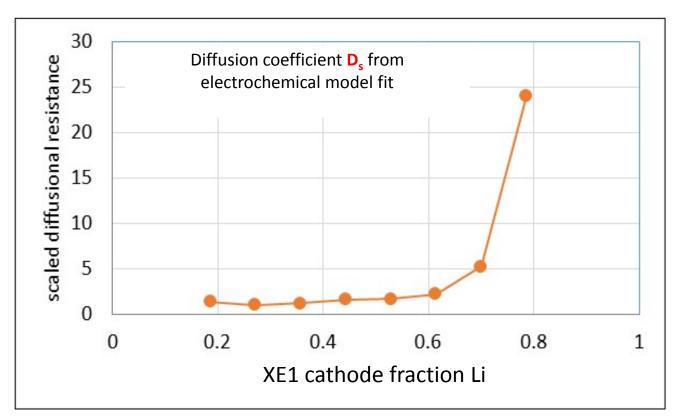
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Solid-State Diffusional Resistance

$$R_d = \frac{r_p^2}{3 F L \varepsilon_s C_{smax} D_s} \left(-\frac{dU}{dy} \right)$$

Daniel Baker and Mark Verbrugge, Journal of the Electrochemical Society, **160** (8), A1319-A1332 (2013)

Particle diffusional resistance trend similar to DC-R, mainly responsible for increased DC-R at low SOC (high Li fraction)





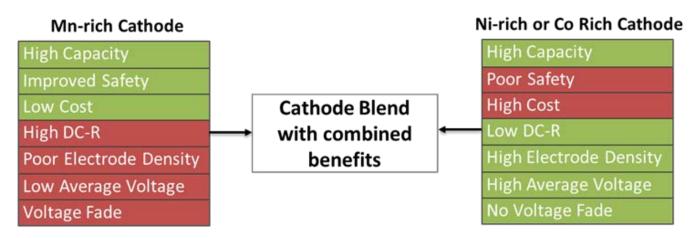






Cathode Blends

Goal: Achieving 200 Wh/kg without using the high voltage capacity from HCMR™-XE



HCMR[™]-XE exhibits low DC-R compared to HCMR[™]-XLE. However, the phase stability or loss of power (increase in DC-R) with cycling is inevitable even in HCMR[™]-XE if operated at higher charge voltages (>4.3V).

Blending HCMR™-XE with other cathode materials benefits with higher active, higher loading in the electrode. This will enable a cell design which allows cell operation at lower charge voltages.

HCMR™-XE structure is maintained showing NO voltage fade (no phase change) when operated in limited upper cut-off window.

Various cathode and anode blends were formulated to meet the ABR cell metrics and cells with the best performing cell design were chosen for the final ABR cell deliverable to INL.

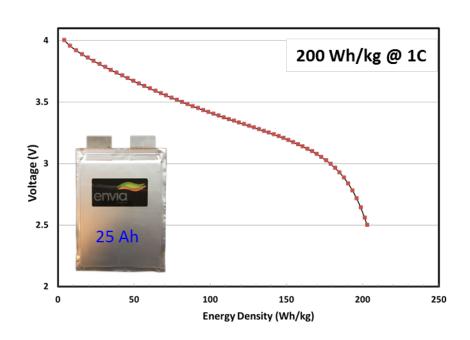


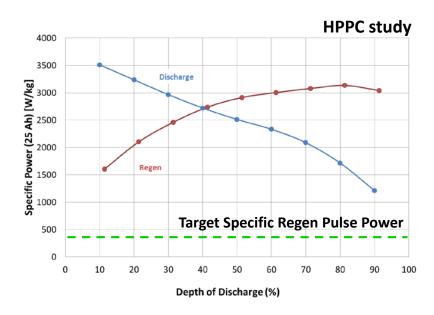






Design 1 – Energy Density and HPPC

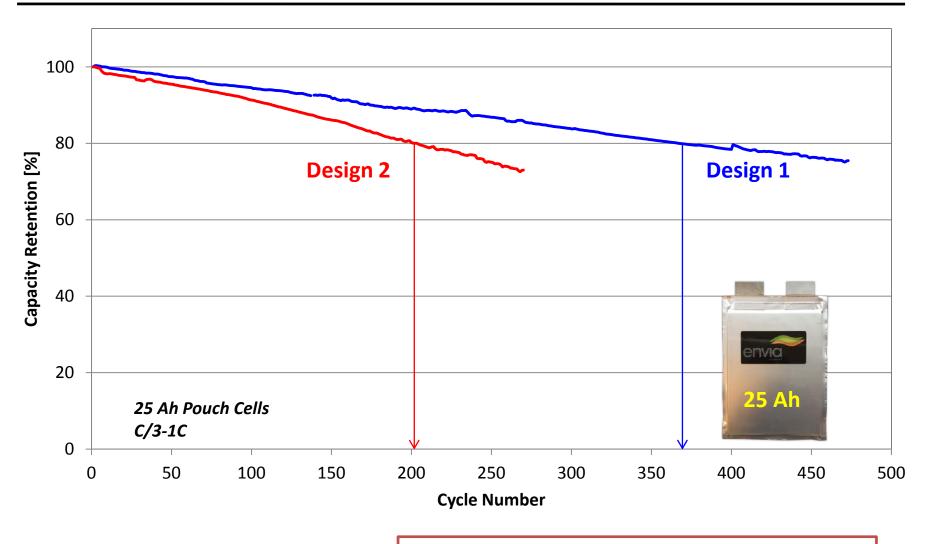




Chracteristics	Unit	PHEV40 ABR Targets	XE Cell after RPT0	
		108010	Design 1	
Specific Discharge Pulse Power	W/kg	800	1718	
Discharge Pulse Power Density	W/L	1600	3693	
Specific Regen Pulse Power	W/kg	430	2103	
Regen Pulse Power density	W/L	860	4523	
Specific Energy @ 1C	Wh/kg	200	203	
Energy Density @ 1C	Wh/L	400	436	
Calendar Life	Years	10+	TBD	
Cycle Life (C/3 ~ 1C)		5000	350	
Operating Temperature	°C	(-30 ~+52)	25 °C	



Designs 1 & 2 – Cycling Performance





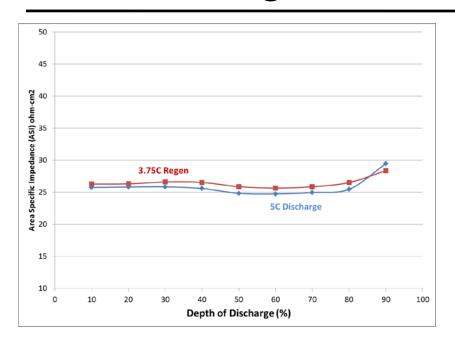


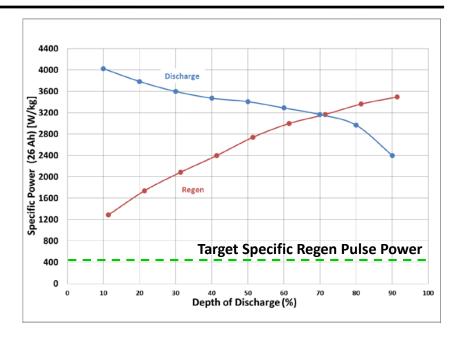




- Met 200 Wh/kg with Designs 1 and 2
- Design 2 has poor cycle-life compared to Design 1
- Design 2 anode has higher Carbon content, but the binder was not optimized for this formulation

Design 4 – HPPC for 26 Ah Cell

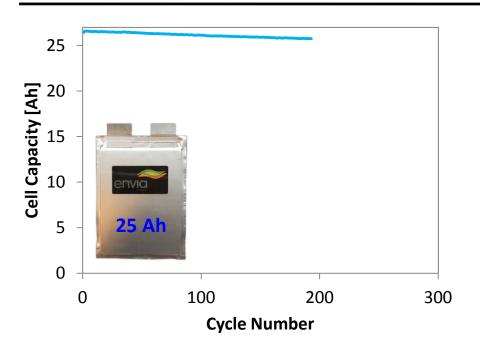


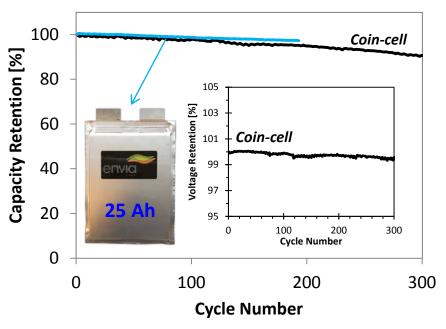


Chracteristics	Unit	PHEV40 - ABR Targets	XE Blen Ce	d: Si-Gr Ils
		largets	Design 1	Design 4
Specific Discharge Pulse Power	W/kg	800	1718	2966
Discharge Pulse Power Density	W/L	1600	3693	6701
Specific Regen Pulse Power	W/kg	430	2103	1739
Regen Pulse Power density	W/L	860	4523	3930
Specific Energy @ 1C	Wh/kg	200	203	210
Energy Density @ 1C	Wh/L	400	436	474
Calendar Life	Years	10+	TBD	TBD
Cycle Life (C/3 ~ 1C)		5000	350	TBD
Operating Temperature	°C	(-30 ~+52)	25 °C	25 °C



Design 4 – Cycling Performance (1)





- Design 4, 25 Ah format cell shows capacity retention >97% after ~190 cycles
- Design 4, 25 Ah format cell is surpassing the analogous
 Design 4 tested in coin-cell format
- Voltage retention remains >99% after 300 cycles in the coin-cell, suggesting no phase-change taking place in the cathode material

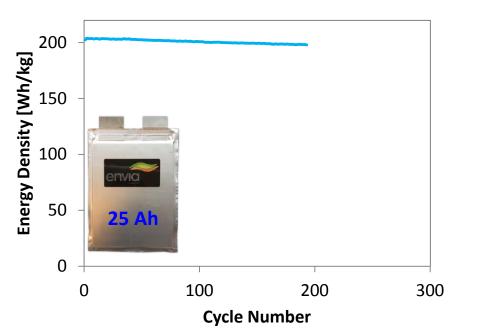


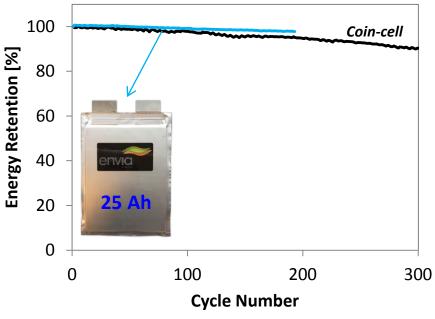






Design 4 – Cycling Performance (2)





- Design 4, 25 Ah format cell shows energy retention >97% after ~190 cycles
- Design 4, 25 Ah format cell is surpassing the analogous Design 4 tested in coin-cell format

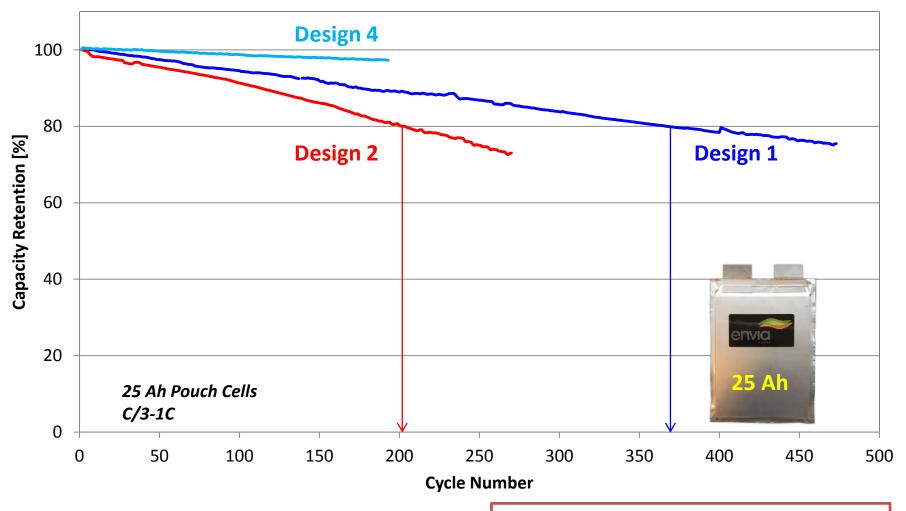








Cycling Performance of Cell Deliverables



- Met 200 Wh/kg @ 1C with Designs 1, 2, and 4
- Design 4 has optimized anode formulation, showing the best cycle ability so far









Conclusions

- HCMR™-XLE cathode showed high capacity, yet very high DC-R which leads to poor power characteristics.
- HCMR™-XLE also showed large increase in resistance with cycling leading to further loss in usable energy.
- Different nanocoatings on HCMR™-XLE such as Al₂O₃, LiPON, polymer etc. via different chemical and physical deposition methods did not reduce DC-R nor did they prevent the increase in DC-R with cycling.
- Composition engineering (from HCMR™-XLE to HCMR™-XE) provided an effective solution for the DC-R challenge for the Li-rich NMC.
- The PFM binder was not compatible with Envia's Si-based anode material resulting in deterioration of the EC characteristics of the anode.
- Cathode blends of HCMR™-XE with Ni-rich or Co-rich cathode materials were developed to operate the cell in the optimized voltage window in order prevent any phase-change driven energy loss with cycling.
- 26 Ah cells were built meeting ABR cell targets of 200 Wh/kg and 400 Wh/L without exposing the cathode to deteriorating voltages.



Acknowledgements

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